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The Elements of Chemical Arithmetic; With a Short System of Elementary Qualitative Analysis (Paperback)

By J Milnor Coit

Rarebooksclub.com, United States, 2012. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book ***** Print on Demand *****.This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1886 Excerpt: .solutions by BaCl₂. Group III. (Hydrochloric Acid Group.) Group reagent, AgNO₃. Hydrochloric, hydrobromic, hydroiodic, hydrocyanic, and hydrosulphuric acids. The acids of this group are precipitated by AgNO₃, and not by BaCl₂. Group IV. (Nitric Acid Group.) Nitric, chloric, and perchloric acids. These acids are not precipitated by any reagent, as all their salts are soluble in water. Reactions of the Inorganic Acids belonging to Group I. 85. Acids precipitated by BaCl₂ in presence of HCl. SULPHURIC ACID, HYDROFLUOSILICIC ACID. Sulphuric Acid, H₂SO₄, 98. 1. BaCl₂ precipitates a white BaSO₄, insoluble in HCl or HNO₃. In very dilute solutions the precipitation is not immediate, but on standing, the solution becomes clouded, and ultimately the precipitate subsides. 2. Pb(NO₃)₂ precipitates a heavy white PbSO₄, soluble in NaHO, and in boiling HCl (on allowing this solution to cool, PbCl₂ crystallizes out). 3. Fused on charcoal with Na₂CO₃ in...



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